

**Example 6.23:** Liquefaction of Nitrogen

Nitrogen is liquefied in the process shown in Figure 6-16. Nitrogen gas is fed at 25 °C, 1 bar (stream 1). The gas is compressed to 180 bar in a compressor with an efficiency of 85%. The precooler uses ambient water as the cooling medium and cools the nitrogen to 30 °C. The precooler uses cold vapor from the separator as the cooling medium; this vapor leaves the precooler at 20 °C (stream). Perform the energy and entropy balances on the basis of 1 kg/s of liquid nitrogen produced.

**Solution** We will use the NIST WebBook to calculate the properties of nitrogen. The pressures of all streams are known: streams 1, 2, 6, 7, 8, and 9 are at 1 bar; streams 3, 4, and 5 are at 180 bar. Streams 6, 7, and 8 is the boiling temperature of nitrogen at 1 bar (−195.91 °C). We notice now that the state of streams 1, 7, 8, and 9 is fully defined and their properties may be collected from the Webbook (see summary table below). In addition to the usual properties ( $P$ ,  $T$ ,  $H$ ,  $S$ , and phase) we must also include in the table of streams the flow rates, since these are not the same among all streams. We set the basis to be 1 kg of liquid nitrogen in stream 7, and the unknown flow rate of vapor stream 8 to be  $x$ . By straightforward material balance, the flow rate of stream 1 is 1 kg/s, the flow rate of streams 8 and 9 are  $x$ , and all other flow rates are  $x + 1$ . The known properties of the streams are summarized below:

	1	2	3	4	5	6	7	8	9
$\dot{m}$	1	$x + 1$	1	$x$	$x$				
$P$	1	1	180	180	180	1	1	1	1
$T$	25			30		195.91	195.91	195.91	25
$H$	309.27			285.15			−122.25	77.073	309.27
$S$	6.8392			5.2159			2.8312	5.411	6.8392

Here,  $\dot{m}$  is the flow rate on the basis of 1 kg of liquid nitrogen,  $P$  is in bar,  $T$  is in °C,  $H$  is kJ/kg, and  $S$  is kJ/kg K.

First we perform an energy balance around a subsystem defined by the cooler, throttling valve, and separator. This part of the process exchanges no heat or work with the surroundings and involves three streams, stream 4 going in and streams 7 and 9 coming out.

*Cooler/throttling/separator.* The subsystem defined by the precooler, the throttling valve, and the separator exchanges no heat or work with the surroundings and it involves 3 streams (4, 7, and 9) whose properties are known. The unknown flow rate  $x$  may be determined by application of the energy balance:

$$x\dot{H}_7 + 1 \times \dot{H}_9 = (1 + x)\dot{H}_4 \quad \Rightarrow \quad x = \frac{\dot{H}_7 - \dot{H}_4}{\dot{H}_4 - \dot{H}_9} = 16.89 \text{ kg/s.}$$

With this, all mass flow rates are now known. Since all vapor in stream 6 exits as stream 8 and all liquid exits as stream 7, the liquid and vapor fractions in stream 6 are

$$x_L = \frac{m_7}{m_7 + m_8} = 0.0559, \quad x_L = 1 - x_V = 0.9441.$$